Haloalkanes and Haloarenes

- 1. S_N1 reaction of alkyl halides lead to
- (a) Retention of configuration
- (b) Racemisation
- (c) Inversion of configuration
- (d) None of these

▼ Answer

Answer: b

- 2. p-djchlorobenzene has higher melting point than its o- and m- isomers because
- (a) p-dichlorobenzene is more polar than o- and m- isomer.
- (b) p-isomer has a symmetrical crystalline structure.
- (c) boiling point of p-isomer is more than o- and m-isomer.
- (d) All of these are correct reasons.

▼ Answer

Answer: b



- 3. Chloropicrin is formed by the reaction of
- (a) steam on carbon tetrachloride.
- (b) nitric acid on chlorobenzene.
- (c) chlorine on picric acid.
- (d) nitric acid on chloroform.

▼ Answer

Answer: d

- 4. Fitting reaction can be used to prepare
- (a) Toluene
- (b) Acetophenon
- (c) Diphenyl
- (d) Chlorobenzene

▼ Answer

Answer: c

5. Identify the end product (C) in the following sequence:

$$C_2H_5OH \xrightarrow{SOCl_2} A \xrightarrow{KCN (alc.)}$$

$$B \xrightarrow{2H_2 O/H^+} C$$

- (a) C₂H₅CH₂NH₂ (b) C₂H₅CONH₂ (c) C₂H₅COOH (d) C₂H₅NH₂ + HCOOH

▼ Answer

Answer: c

6.

$$CH_3CH_2CH_2C1 \xrightarrow{alc. KOH} B \xrightarrow{HBr} C \xrightarrow{Na/ether} D$$

In the above reaction, the product D is

- (a) Propane
- (b) 2, 3-Dimethylbutane
- (c) Hexane
- (d) Allyl bromide
- **▼** Answer





7. Identify X and Y in the following sequence

 C_2H_5 Br \xrightarrow{X} Product \xrightarrow{Y} $C_3H_7NH_2$

- (a) X = KCN, $Y = LiAlH_4$
- (b) $X = KCN, Y = H_3O^+$
- (c) $X = CH_3Cl$, $Y = AlCl_3HCl$
- (d) $X = CH_3NH_2$, $Y = HNO_2$

▼ Answer

Answer: a

8. In the following sequence of reactions:

 $C_2H_5Br \xrightarrow{AgCN} X \xrightarrow{Reduction} Y; Y is$

- (a) n-propylamine
- (b) isopropylamine
- (c) ethylamine
- (d) ethylmethylamine

▼ Answer

Answer: d

9.

 $X \xrightarrow{\text{AgNO}_3} Yellow \text{ or While ppt}$

Which of the following cannot be X?

(d)
$$\bigcirc$$
 N_2^{\uparrow} CI

▼ Answer

Answer: a



10.

Identifay Z in the series

$$CH_2 = CH_2 \xrightarrow{HBr} X \xrightarrow{aq. KOH} Y$$

$$\xrightarrow{Na_2CO_3} Z$$

$$I_2 \text{ excess} Z$$

- (a) C₂H₅I
- (b) C_2H_5OH
- (c) CHI₃
- (d) CH₃CHO
- **▼** Answer

Answer: c

